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Hydrophobicity
characterization and binding
studies of non-aqueous medium
hydrophobic macromolecules
to *Vitis amurensis* agglutinin II.
The nonspecific
macromolecular agglutination
reaction between carbohydrates
and proteins and carbohydrates
and lectins is known to occur in
hydrophobic systems. However,
the nature of the interaction
between *Vitis amurensis*

agglutinin II (VAAII) and hydrophobic macromolecules, such as polystyrene latexes, dextran, dextran sulfate (DS), polyethylene glycol (PEG) of different molecular weight, and tannin was studied in aqueous and non-aqueous solution. By fluorescence quenching assay, the maximum binding constants for VAAII and PEG in aqueous solution were calculated to be 1.78×10^4 and 2.42×10^4

L/mol, respectively. The maximum binding constants for the interaction of VAAII and PEG in hexadecane, isooctane, tricaprylin, and tannin were calculated to be $3.00 \times 10(3)$, $1.29 \times 10(3)$, $3.41 \times 10(2)$, and $1.20 \times 10(2)$ L/mol, respectively. VAAII was found to have no affinity to dextran, dextran sulfate, polyethylene glycol in hexadecane, isooctane, and tricaprylin; however, the

biding constants to PEG in these two solvents were calculated to be 2.96×10^4 and 1.79×10^4 L/mol. Hence, it is concluded that there was a hydrophobic interaction between VAAII and hydrophobic macromolecules such as polystyrene latexes, dextran, dextran sulfate, polyethylene glycol, and tannin in non-aqueous medium. The results were discussed with

respect to the molecular motion of the macromolecules in non-aqueous medium. NASA is the latest recipient of the U.S. National Medal of Technology for technology that has helped transform humanity's relationship with the sun. The agency's work in solar power and solar cells have made solar energy more energy efficient and opened up new possibilities for space exploration. In

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